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Key indicators

Single-crystal X-ray study
 T = 293 K
 Mean $\sigma(C-C)$ = 0.004 Å
 R factor = 0.060
 wR factor = 0.132
 Data-to-parameter ratio = 16.2

For details of how these key indicators were
 automatically derived from the article, see
<http://journals.iucr.org/e>.

cis-Bis(dicyanamido)bis(1,10-phenanthroline)-
 cadmium(II)

In the title complex, $[Cd(C_2N_3)_2(C_{12}H_8N_2)_2]$, the Cd^{II} atom exhibits a slightly distorted octahedral environment, coordinated by four N atoms of two phenanthroline ligands and two mutually *cis* N-terminal atoms of two dicyanamide ligands. π - π stacking interactions result in the formation of one-dimensional chains extending along [110].

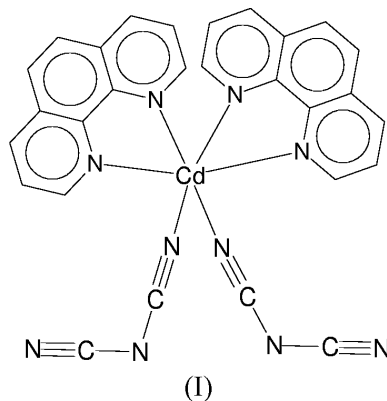
Comment

Investigation of metal-dca complexes {dca is dicyanamide, $[N(CN)_2]^-$ } is a fast-growing research field because of the large variety of topologies and magnetic properties shown by complexes of the dca ligand (Miller & Manson, 2001; Batten & Murray, 2003). It is a versatile ligand coordinating to metal ions in various modes: monodentate binding through one nitrile N atom (Marshall *et al.*, 2000), end-to-end bridging through the two nitrile N atoms (Manson, Arif & Miller, 1999; Jensen *et al.*, 1999) and triply bridging three metal atoms using all three N donor atoms (Jensen *et al.*, 2000; Kurmco & Kepert, 1998). These varied coordination possibilities allow the preparation of compounds with a variety of architectures, including mono- and dinuclear complexes, as well as one-, two- and three-dimensional network structures. Complexes $[M(dca)_2]_n$ ($M = Mn, Fe, Co, Ni, Cu, Zn$ and Ag) containing only dca have been synthesized, and display quite limited structural types with rutile-like or two-dimensional (4,4) sheet networks (Manson *et al.*, 1998; Batten *et al.*, 1998). By introducing co-ligands of various types, such as pyridine, bipyridine, 1,10-phenanthroline or 2,2'-biimidazole, many structurally diverse ternary compounds have been synthesized (Marshall *et al.*, 2000; Manson, Arif, Incarvito *et al.*, 1999; Potocnak *et al.*, 1995). As an extension of this research, we report here the synthesis and structure of the title compound, (I).

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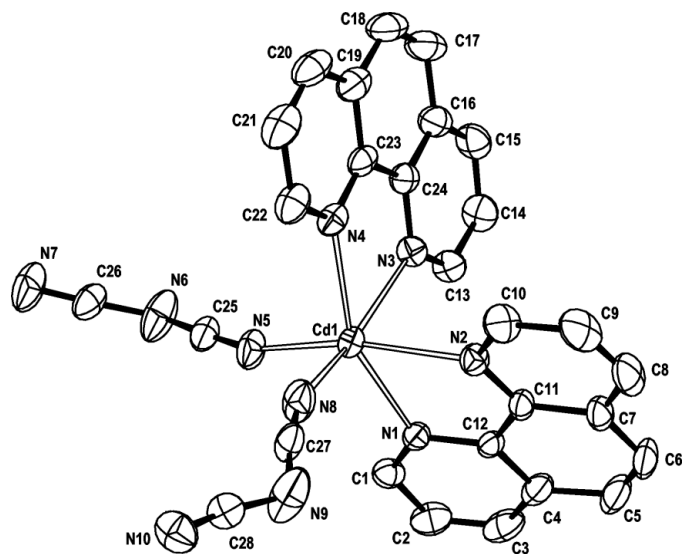


Figure 1

The molecular structure of the title compound, with displacement ellipsoids drawn at the 30% probability level; H atoms have been omitted for clarity.

The structure of (I) consists of discrete molecules (Fig. 1) and is similar to those of the complexes $M(\text{dca})_2(1,10\text{-phen})_2$ ($M = \text{Cu}, \text{Mn}, \text{Zn}, \text{and Ni}$) (Wu *et al.*, 2004; Potocnak *et al.*, 1995; Wang *et al.*, 2000). The Cd^{II} atom is in a slightly distorted octahedral environment, coordinated by four N atoms from the two phen ligands and two N-terminal atoms from two dca ligands. The two dca ligands bind to Cd in a monodentate fashion through their nitrile N atoms, N5 and N8, and are mutually *cis*. The bond lengths of the Cd–N bonds to the phen ligands [mean 2.365 (2) Å] are distinctly longer than those to the nitrile N atoms of dca [mean 2.254 (2) Å]. This difference may be due to steric hindrance involving the bulky phen molecules. The structure is stabilized by weak π – π stacking interactions between phen ligands from adjacent molecules of (I) to form a one-dimensional chain structure extending along [110]. Adjacent rings are inclined at an angle of 1.8 (1)°; the ring centroid-to-centroid distance is 3.816 Å and the perpendicular inter-ring distance is 3.807 Å (Fig. 2). The shortest $\text{Cd}^{\text{II}} \cdots \text{Cd}^{\text{II}}$ separation along the chain is 8.726 (2) Å. Rings R1 (N1/C1–C4/C12) interact with rings R2 (C4–C7/C11/C12) on adjacent molecules to give a single one-dimensional chain. This interaction differs markedly from that reported for the coordination polymer formed by the copper complex of phen and a bridging dca ligand (Wu *et al.*, 2003), in which an interleaved double-chain structure was formed.

Experimental

An ethanol solution (10 ml) of 1,10-phen (100 mg, 0.51 mg) was added dropwise, with stirring, to an aqueous solution (6 ml) of $\text{Cd}(\text{CH}_3\text{COO})_2$ (130 mg, 0.49 mmol) and Na(dca) (88 mg, 0.99 mmol). The mixture was filtered and the filtrate was allowed to stand for several days, yielding transparent colourless block-like crystals (yield 63.5%, based on Cd).

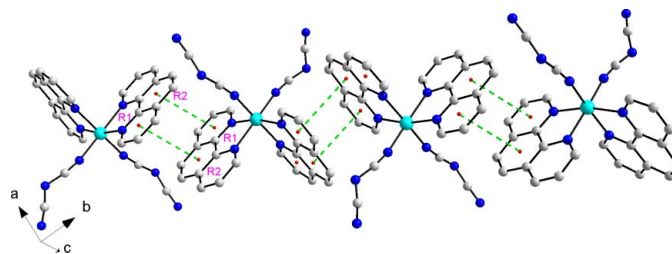


Figure 2

The chain structure of (I), resulting from π – π stacking interactions (dashed lines), extending along [110].

Crystal data

$[\text{Cd}(\text{C}_2\text{N}_3)_2(\text{C}_{12}\text{H}_8\text{N}_2)_2]$
 $M_r = 604.91$
 Monoclinic, $P2_1/c$
 $a = 9.878$ (2) Å
 $b = 14.978$ (4) Å
 $c = 17.764$ (5) Å
 $\beta = 104.691$ (3)°
 $V = 2542.2$ (11) Å³
 $Z = 4$

$D_x = 1.580$ Mg m^{−3}
 Mo $K\alpha$ radiation
 Cell parameters from 5717 reflections
 $\theta = 3.0$ – 27.5 °
 $\mu = 0.90$ mm^{−1}
 $T = 293$ (2) K
 Block, colourless
 $0.20 \times 0.15 \times 0.10$ mm

Data collection

Rigaku Mercury CCD diffractometer
 ω scans
 Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2002)
 $T_{\text{min}} = 0.777$, $T_{\text{max}} = 0.914$
 19 784 measured reflections

5721 independent reflections
 4856 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$
 $\theta_{\text{max}} = 27.5$ °
 $h = -11 \rightarrow 12$
 $k = -14 \rightarrow 19$
 $l = -23 \rightarrow 23$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.060$
 $wR(F^2) = 0.133$
 $S = 1.07$
 5721 reflections
 353 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0471P)^2 + 4.8527P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.79$ e Å^{−3}
 $\Delta\rho_{\text{min}} = -0.61$ e Å^{−3}
 Extinction correction: none

Table 1

Selected geometric parameters (Å, °).

| | | | |
|-----------|-------------|----------------------|-------------|
| Cd1–N8 | 2.251 (2) | Cd1–N2 | 2.373 (2) |
| Cd1–N5 | 2.257 (2) | Cd1–N3 | 2.4040 (19) |
| Cd1–N4 | 2.335 (2) | Cd1–Cd1 ⁱ | 8.7257 (15) |
| Cd1–N1 | 2.3484 (17) | | |
| N8–Cd1–N5 | 98.47 (9) | N4–Cd1–N2 | 94.61 (7) |
| N8–Cd1–N4 | 93.11 (8) | N1–Cd1–N2 | 70.69 (7) |
| N5–Cd1–N4 | 100.32 (8) | N8–Cd1–N3 | 161.88 (8) |
| N8–Cd1–N1 | 105.59 (8) | N5–Cd1–N3 | 92.27 (8) |
| N5–Cd1–N1 | 92.65 (7) | N4–Cd1–N3 | 70.53 (6) |
| N4–Cd1–N1 | 155.44 (6) | N1–Cd1–N3 | 88.33 (6) |
| N8–Cd1–N2 | 88.22 (8) | N2–Cd1–N3 | 85.50 (7) |
| N5–Cd1–N2 | 163.23 (7) | | |

Symmetry code: (i) $-x + 1, -y + 2, -z$.

H atoms were placed in idealized positions and refined as riding, with C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *CrystalClear* (Rigaku, 2002); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXTL* (Siemens, 1994); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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